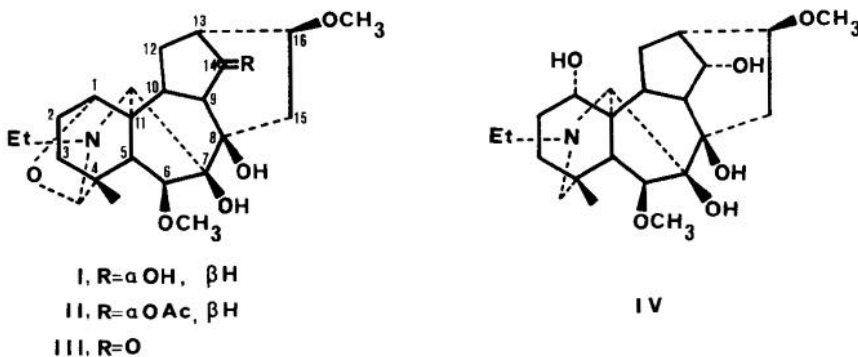


GADESINE, A NEW C-19 DITERPENE ALKALOID FROM DELPHINIUM PENTAGYNUM LAM

A G González G de la Fuente and R Díaz
Instituto de Productos Naturales Orgánicos CSIC
La Laguna Tenerife SPAIN

&
J Fayos and M Martínez-Ripoll
Dpto de Rayos-X Instituto Rocasolano Serrano 119 Madrid 6 SPAIN

We should like to report the structure determination of gadesine (I), a C-19 diterpene alkaloid isolated from the plant Delphinium pentagynum Lam, collected in Cadiz (Southern Spain).



Gadesine, $C_{23}H_{35}NO_6$: mp 174-177°, $\{\alpha\}_D^{+76}$ (c 0.27, EtOH); IR (KBr), cm^{-1} 3520, 3400 (OH), 1125, 1090 (ether linkage), 1000 and 900 (carbinolamine inner ether¹); PMR ($CDCl_3$), δ 1.00 (3H, s, C-CH₃), 1.09 (3H, t, J=7Hz, N-CH₂-CH₃), 3.35, 3.37 (3H each, s, OCH₃), 3.69 (1H, m, $W_{1/2}$ =7Hz, C-1 β H), 3.78 (1H, s, C-19H), 3.82 (1H, d, J=1Hz, C-6 α H), 3.97 (1H, dd, $J_1=J_2=4.5$ Hz, C-14 β H) and 2.99, 3.91 and 3.92 which disappear when D₂O is added (1H each, s, OH). The MS is characteristic of alkaloids with lycocotonine skeletons: M^+ 421 (4%), M^+ -CH₃ (82%, loss of a methyl group from N-ethyl), $\{M^+$ -CH₃\}-H₂O (base peak), M^+ -CH₃OH (24%) and M^+ -56 (12%, ejection of acrolein from ring A owing to the carbinolamine inner ether)².

Treated with Ac_2O/Py , gadesine gave monoacetate II as a resin: M^+ 463; IR (KBr), cm^{-1} 3450 (OH), 1730 and 1250 (acetate); PMR ($CDCl_3$), δ 2.08 (3H, s, OCOCH₃) and 4.84 (1H, dd, J=4.5Hz, C-14 β H). Oxidation of I with Cornforth's reagent yielded 14-ketogadesine (III): mp 157-160°, $\{\alpha\}_D^{+51}$ (c 0.2, EtOH); M^+ 419; IR (KBr), cm^{-1} 3450 (OH) and 1745 (pentacyclic ketone). $LiAlH_4$ reduction of I gave the amino-alcohol IV: mp 80-82°, $\{\alpha\}_D^{+32}$ (c 0.1, EtOH); IR (KBr), cm^{-1} 3400 and 3120 (OH) did not show the absorption proper to the carbinolamine inner ether; PMR ($CDCl_3$), δ 3.73 (1H, m, $W_{1/2}$ =6Hz, C-1 β H, boat conformation in ring A)³; MS, M^+ 423 (21%), M^+ -OH (61%, loss of C-1 α OH). This spectrum was very like those of delsoline, delcosine and acetyl delcosine⁴ underlining the structural

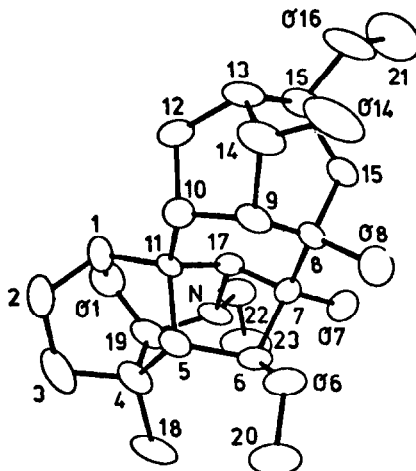
similarity of these bases. IV was converted in 75% yield to gadesine (mp, IR, PMR, MS) by KMnO_4 oxidation⁵, thus confirming the presence of C-1 α OH in IV and of the C-1-C-9 ether in gadesine.

The above chemical and spectroscopic data and biogenetic considerations led to structure I being tentatively assigned to gadesine. In order to confirm this structure an X ray analysis was performed. Crystal data: monoclinic space group $P2_1$ with two molecules of I per unit cell, dimensions $a=12.320(3)$, $b=8.652(1)$, $c=10.148(2)\text{\AA}$ and $\beta=104.72(1)^\circ$. 2702 reflections within the range $2<\theta<28^\circ$ were recorded on a 4-circle diffractometer with monochromated $\text{MoK}\alpha$ radiation ($\lambda=0.7107\text{\AA}$). There was 39% intensity decay in 50h experiment so two crystals were used. 1963 reflections were considered observed $\{I>2\sigma(I)\}$. The crystal structure was solved by direct methods and refined anisotropically (isotropically for H atoms), with unit weight, by least-squares analysis to $R=7.8\%$ and $R_w=7.5\%$ ⁶.

Ring A is stabilized in the skew form through the ether bridge C1-O1-C19, contributing to a very rigid unit, C1-C2-C3-C4-C19-N-C17-C11, bridged twice through O1 and C5. Two intramolecular H bonds are formed between O8-H8...O6 (2.68\AA) and O14-H14...O16 (3.11\AA).

Gadesine is an interesting alkaloid from the biogenetic point of view, being, like delcosine, 14-dehydrodelcosine, mono-acetyldelcosine⁷ and delcorine⁸ a lycocotnine-type alkaloid with C-1 α and C-6 β substituents.

FINAL X-RAY MODEL FOR GADESINE



ACKNOWLEDGEMENT We are grateful to Prof S García-Blanco for the use of his Dept's automatic diffractometer and to Dr J Rivera for taking the high resolution MS.

R E F E R E N C E S

- 1 R Auct D Clayton and L Marion *Can J Chem* **35** 397 (1957)
- 2 S W Pelletier and S W Page "Chemistry of the Diterpene Alkaloids" in "The Alkaloids" (Spec Per Rep) ed: J E Saxton The Chem Soc London 1973 vol 3 p235
- 3 S W Pelletier Z Djarmati S Lajsić and W H De Camp *J Am Chem Soc* **99** 2617 (1976)
- 4 S D Sastry "Biochemical Applications of Mass Spectrometry" ed: G R Waller J Wiley 1972 p662
- 5 O Achatowicz Y Tsuda and L Marion *Can J Chem* **43** 2336 (1965)
- 6 Crystallographic programme used: P Main M M Woolfson J P Declercq and G Germain "Multan System" Physics Dept Univ of York England 1974; J M Stewart F A Kundell and J C Baldwin "The X-Ray System of Crystallographic Programs" Computer Science Center Univ Maryland USA 1970
- 7 S W Pelletier and L H Keith "The Alkaloids" ed: R H F Manske Academic Press 1970 Vol XII p10
- 8 A S Narzullaev M S Yunusov and S Yu Yunusov *Khim Prir Soedin* **9** 497 (1973)

(Received in UK 31 October 1978)